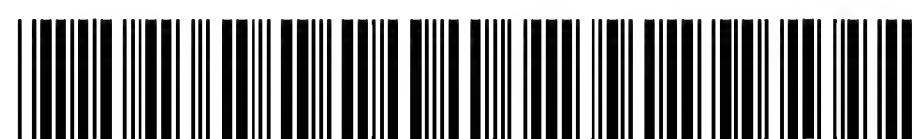


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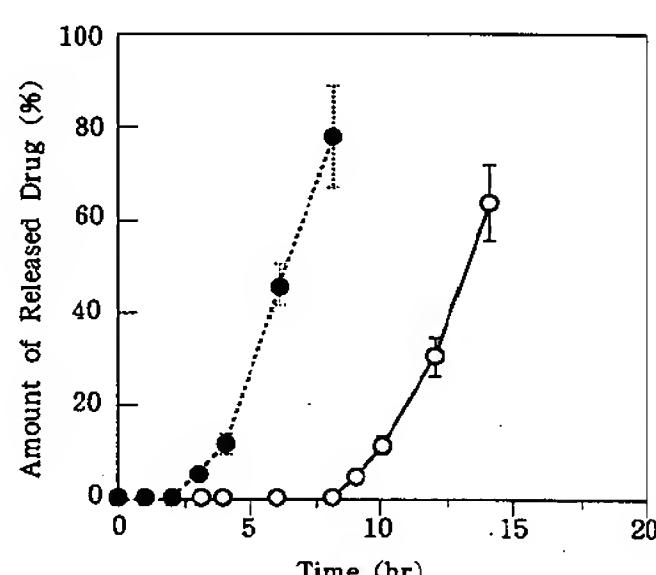
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**㉕ PERORAL PHARMACEUTICAL PREPARATION RELEASABLE IN LOWER DIGESTIVE TRACT.**

㉖ A peroral pharmaceutical preparation releasable in the lower digestive tract, having a double-layer coating structure wherein a solid drug comprising a core containing an active ingredient is coated with an inner layer comprising a cationic polymer and an outer layer comprising an anionic polymer.

FIG. 1



**EP 0 667 148 A1**

## Technical Field

5 The present invention relates to an oral pharmaceutical preparation which is releasable in the lower digestive tract. In particular, the present invention relates to a pharmaceutical preparation for oral administration which passes through the stomach without being affected by gastric juice, does not undergo any change in the small intestine, and begins to disintegrate for the first time when it arrives at the large intestine to release the active ingredient, which is absorbed therein.

## Background Art

10 The pharmaceutical preparations for oral administration according to the prior art begin to disintegrate or dissolve before the arrival at the large intestine. Therefore, it has been a practice in the prior art to employ an intravenous, intranasal or rectal administration method in the administration of an active substance which is expected to exhibit a local action on the large intestine or is decomposed by the lyase present in the small intestine.

15 However, an oral pharmaceutical preparation has the following advantages. Namely, if an oral pharmaceutical preparation releases a pharmacologically active substance selectively in the large intestine, the concentration of the active substance at the release site can be increased to give a more effective local action than that of the preparation according to the prior art. Further, if a pharmacologically active protein or 20 peptide which is decomposed in the small intestine to lose its activity is released selectively in the large intestine, the protein or peptide will be able to be absorbed in the large intestine wherein the protease activity is poor. Owing to the above advantages, it is apparent that an oral pharmaceutical preparation is a dosage form far more useful than injection or suppository.

25 For the reason described above, many attempts have recently been made to develop oral pharmaceutical preparations targeting the large intestine.

30 For example, there have been reported an oral preparation targeting the large intestine, which is prepared by combining a polymer soluble only at a pH of 5.5 or above with an insoluble polymer (EP40590); a solid oral preparation coated with a suitable amount of an anionic polymer soluble at a pH of 7.0 or above (trade name: Eudragit S, a product of RÖHM) (WO83/00435); an oral preparation coated with a 35 mixture comprising an anionic polymer soluble at a pH of 7.0 or above (trade name: Eudragit S or L, a product of RÖHM) and a methacrylate copolymer difficultly soluble in water (trade name: Eudragit RS, a product of RÖHM) at a suitable ratio (EP225189); an osmotic-pump preparation coated with an enteric polymer (BE903502); a colon-reaching oral preparation covered with an inner coat soluble at a pH of 7.0 or above, an intermediate coat made of a gelatinized polymer, and a acid-resistant outer coat soluble at a pH of 5.5 or above (Japanese Patent Publication No. 501411/1992); and so forth.

40 All of these preparations are characterized in that a drug is released by a pH increase in the intestinal tract, and they are each prepared by applying a proper amount of an anionic polymer or by adding or applying an insoluble polymer to prevent the drug from being released during its residence in the small intestine.

45 Meanwhile, a recent study on humans (D.F. Evans, Gut, 1988, vol.29, p.1035) reported that the pH of the intermediate and lower parts of the small intestine ranges from 7.4 to 7.5 (standard deviation: 0.4 to 0.5), while the average pH rapidly drops to 6.4 (standard deviation: 0.4) in the upper part of the large intestine and gradually increases to reach 7.0 (standard deviation: 0.7) in the lower part of the large intestine.

50 However, all of the above oral preparations releasable in the large intestine according to the prior art are coated with a polymer soluble or swelling at a pH of 7.0 or above, on the assumption that the pH in the colon or large intestine is 7.0 or above. The inventors of the present invention have also made intensive studies on this respect and have found that the pH of cecal contents of a rat or rabbit is below the range of 6.4 to 5.5 owing to the organic acids generated by enteric bacteria. This result agrees with that disclosed in the above literature on humans. Accordingly, the inventors of the present invention have started studies to develop an oral pharmaceutical preparation from which a drug is released in the lower digestive tract (large intestine) by a pH change to below the range of 7.0 to 5.5.

55 Further, WO90/13286 discloses an oral preparation comprising a core containing a basis and a filler, a first layer which covers the core and is composed of an enteric component, a second layer which covers the first layer and is composed of a nonenteric component, and a third layer which covers the second layer and is composed of an enteric component. In this oral preparation, the second layer, which is composed of a cationic polymer, is covered with the third layer to regulate the release of a drug, so that it is very difficult to regulate the preparation so as to release a drug specifically in the large intestine. Further, since the core

always contains an acidic substance, the second layer is liable to be dissolved in the small intestine to release a drug prior to the arrival at the large intestine.

The coat according to the prior art has such a property that the dissolution or swelling thereof proceeds before the arrival of the preparation at the large intestine. Therefore, the preparation of the prior art has a 5 disadvantage that when the actual residence time in the small intestine, which has a great physiological influence, is shorter than the one assumed in the dosage form design, the preparation is discharged as such, while when the actual residence time is longer than the assumed one, the release of a drug has been completed before the arrival at the target site.

10 Disclosure of the Invention

The present invention aims at solving the above problem of the prior art to develop an oral pharmaceutical preparation which passes through the stomach without being affected by gastric juice, does not undergo any change in the small intestine, and begins to disintegrated for the first time when it arrives 15 at the large intestine to release the active ingredient, which is absorbed therein.

The oral pharmaceutical preparation releasable in the lower digestive tract according to the present invention is characterized by having a double-coated structure wherein a solid drug having a core containing an active ingredient is covered with both an inner coat made of a cationic polymer and an outer coat made of an anionic polymer.

20 Namely, the present invention relates to a novel oral pharmaceutical preparation which comprises a solid drug having a core containing a therapeutically active ingredient which is releasable in the lower digestive tract, and the following coats (①) and (②):

coat (①): inner coat which contains a suitable plasticizer or binding inhibitor and is made of a cationic polymer soluble or swelling at a pH of 6.0 or below, and

25 coat (②): outer coat made of an anionic polymer soluble at a pH of 5.5 or above.

The solid drug having the core is made of a therapeutically active ingredient alone or a mixture thereof with a filler, binder, disintegrator and/or sorbefacient.

Examples of the filler include lactose, crystalline cellulose, crystalline cellulose sodium carboxymethyl-cellulose, wheat starch, and magnesium stearate.

30 Examples of the binder include crystalline cellulose, pullulan, acacia, sodium alginate, polyvinyl pyrrolidone and macrogol.

Examples of the disintegrator include ordinary inert substances for pharmaceutical preparations, such as carboxymethylcellulose, calcium carboxymethylcellulose, hydroxypropylcellulose, hydroxypropylstarch, starch and sodium alginate.

35 Further, a sorbefacient may be used in the preparation of the solid drug to accelerate the absorption of the active ingredient, and examples of such a sorbefacient include sugar ester, sucrose esters of fatty acid, glycyllsinate salts, glycyrrhetic acid, dipotassium glycyrrhizinate, bile acid, glycerol esters of fatty acid, 1-[(2-(decylthio)ethyl]azacyclopentan-2-one, adipic acid, basic amino acids, polyethylene glycol and sodium caprate.

40 The solid drug may be encapsulated with gelatin or may be granulated or tabletted.

The therapeutically active ingredient to be contained in the solid drug is not particularly limited but may be any substance which exhibits its effect when released in the lower digestive tract. Examples of the therapeutically active ingredient include peptides, proteins, anti-inflammatory agents, antineoplastic agents, antibiotics, chemotherapeutics, ulcerative colitis remedy, irritable colitis remedy (choline blocker), and 45 constipation remedy. Specific examples of the therapeutically active ingredient include somatostatin, insulin, calcitonin, vasopressin, gastrin, EGF (epidermal growth factor),  $\alpha$ -hANP ( $\alpha$ -human atrial natriuretic peptide), enkephalin, endorphin, GM-CSF (granulocyte macrophage colony-stimulating factor), G-CSF (granulocyte colony-stimulating factor), human growth hormone, glucagon, t-PA (tissue plasminogen activator), TNF (tumor necrosis factor), TCGF (T cell growth factor), ACTH (adrenocorticotropic hormone), interleukins, 50 interferon, EPO (erythropoietin), urokinase, neocarcinostatin, bradykinin, immunoglobulin and its digestion product, various allergens and their digestion products, ketoprofen, ibuprofen, diclofenac, indometacin, ketrolac, fenbufen, loxoprofen, tenidap, piroxicam, tenoxicam, salazosulfapyridine, pipethanate hydrochloride, mepenzolate bromide, and sennosides A and B.

The cationic polymer which is soluble or swelling at a pH of 6.0 or below and is used in the formation of 55 the inner coat is an aminoalkyl methacrylate copolymer (a copolymer comprising methyl methacrylate, butyl methacrylate and dimethylaminoethyl methacrylate, trade name: Eudragit E, a product of RÖHM) or a polyvinyl acetal diethylaminoacetate (trade name: AEA, a product of Sankyo). The inner coat is formed by applying 1 to 40% by weight (based on the solid drug) of the above polymer to the solid drug in a film

thickness of 10 to 300  $\mu\text{m}$ , by which the resulting preparation is regulated so as to release the active ingredient therefrom speedily when the pH condition of 6.0 or below continues.

It is preferable that a suitable plasticizer be used in the formation of the inner coat to make the coat smooth. The plasticizer includes triacetin, citrate ester and polyethylene glycol. The binding inhibitor 5 includes talc, titanium oxide, calcium phosphate, hydrophobic light anhydrous silicic acid, and so forth.

The outer coat is made of an anionic polymer which is easily soluble at a pH of 5.5 or above. Examples of the anionic polymer usable in the present invention include methacrylic acid copolymer L (a copolymer comprising methacrylic acid and methyl methacrylate, trade name: Eudragit L100, a product of RÖHM), methacrylic acid copolymer S (a copolymer comprising methacrylic acid and methyl methacrylate, trade 10 name: Eudragit S, a product of RÖHM), hydroxypropylcellulose, hydroxyethylcellulose phthalate, polyvinyl acetate phthalate, cellulose acetate phthalate, and cellulose acetate tetrahydrophthalate. The anionic polymer is applied in an amount of 1 to 40% by weight based on the solid drug.

#### Brief Description of Drawings

- 15 Fig. 1 is a graph showing the results of in vitro test on the release of drug.  
 Fig. 2 is a graph showing the results of in vivo test on blood drug concentration.  
 Fig. 3 is a graph showing the influence of coating weight on the release of drug in vitro.  
 Fig. 4 is a graph showing the influence of pH on the release of ketoprofen.  
 20 Fig. 5 is a graph showing the influence of pH on the release of FITC dextran.  
 Fig. 6 is a graph showing the influence of pH on the release of calcitonin.  
 Fig. 7 is a graph showing the results of the blood calcium concentration test using dog.  
 Fig. 8 is a graph showing the influence of coating weight on the release of drug in vitro.  
 Fig. 9 is a graph showing the effect of sorbefacient in vivo.

#### Best Mode for Carrying Out the Invention

The present invention will now be specifically described by referring to the following Examples.

#### 30 Example 1

A ketoprofen-containing solid drug was prepared according to the following formula:

35	ketoprofen magnesium stearate lactose	24.0 pt. by wt. 1.0 pt. by wt. 75.0 pt. by wt.
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40 Ketoprofen, lactose and magnesium stearate were mixed together to form a homogeneous mixture. This mixture was compressed with a tablet machine into tablets having a diameter of 7 mm and a weight of 210 mg. The obtained solid drug was coated as follows.

45	Eudragit E (trade name) ethanol water talc	7.0 pt. by wt. 70.0 pt. by wt. 19.5 pt. by wt. 3.5 pt. by wt.
----	---	--

50 A solution comprising the above components was continuously sprayed at 50 °C on the solid drug preheated to 50 °C to form an inner coat. The weight increase of the solid drug was 30 mg.

After the completion of the spraying, the resulting drug was dried and further coated as follows.

5

Eudragit S (trade name)	7.0 pt. by wt.
ethanol	70.0 pt. by wt.
water	18.8 pt. by wt.
talc	3.5 pt. by wt.
polyethylene glycol 6000	0.7 pt. by wt.

A solution comprising the above components was continuously sprayed at 50 °C on the resulting drug preheated to 50 °C to form an outer coat. The weight increase of the solid drug was 20 mg.

10

Comparative Example 1

A ketoprofen-containing solid drug was prepared according to the following formula:

15

ketoprofen	24.0 pt. by wt.
magnesium stearate	1.0 pt. by wt.
lactose	75.0 pt. by wt.

20 Ketoprofen, lactose and magnesium stearate were mixed together to form a homogeneous mixture. This mixture was compressed with a tablet machine into tablets having a diameter of 7 mm and a weight of 210 mg. The obtained solid drug was coated as follows.

25

Eudragit S (trade name)	7.0 pt. by wt.
ethanol	70.0 pt. by wt.
water	18.8 pt. by wt.
talc	3.5 pt. by wt.
polyethylene glycol 6000	0.7 pt. by wt.

30

A solution comprising the above components was continuously sprayed at room temperature on the solid drug preheated to 50 °C to form an inner coat. The weight increase of the solid drug was 30 mg. The resulting drug was dried and further coated as follows.

35

hydroxypropylmethylcellulose	4.0 pt. by wt.
polyethylene glycol 400	0.5 pt. by wt.
95% ethanol	85.5 pt. by wt.
water	10.0 pt. by wt.

40

Hydroxypropylcellulose was suspended in 95% ethanol, followed by the addition of a solution of polyethylene glycol 400. The obtained solution was sprayed on the above solid drug preheated to 50 °C to form an intermediate coat. The weight increase of the solid drug was 10 mg.

45 The resulting solid drug was furthermore coated by spraying a solution comprising the following components like in the formation of the above coat. The weight increase of the solid drug was 10 mg.

50

Eudragit L100 (trade name)	7.0 pt. by wt.
ethanol	70.0 pt. by wt.
water	18.8 pt. by wt.
talc	3.5 pt. by wt.
polyethylene glycol 6000	0.7 pt. by wt.

55 (A) In vitro test on the release of drug

The tablets prepared in Example 1 and Comparative Example 1 were examined for the release of ketoprofen. Each tablet was incubated in a buffer of pH2.0 for 2 hours, in a buffer of pH7.4 for 6 hours, and

then in a cecal solution prepared by suspending the cecal contents in a phosphate buffer of pH7.4 containing 125 mg of glucose and 4.1 mg of benzyl viologen. The amount of the released ketoprofen was determined by liquid chromatography. The results are given in Fig. 1, wherein ○ represents the release pattern of the tablet prepared in Example 1 and, ● represents that of the tablet prepared in Comparative Example 1.

As apparent from the results of Fig. 1, the release of ketoprofen from the tablet prepared in Example 1 was observed specifically in a cecal solution, though it was not observed in a buffer of pH2.0 or 7.4 at all. On the other hand, most of the ketoprofen contained in the tablet prepared in Comparative Example 1 was released in a buffer of pH7.4.

10 (B) In vivo test on blood drug concentration

The tablet prepared in Example 1 was administered to a laboratory animal (beagle dog, male, weight: 10 kg) without anesthesia. The result is given in Fig. 2, wherein ● represents the result of the tablet of Example 1.

15 As apparent from the result of Fig. 2, no ketoprofen was observed in the plasma in the early stages, though ketoprofen is very readily absorbable. After 14 hours and later, however, ketoprofen was observed in the plasma at extremely high concentrations.

20 Example 2

Five kinds of tablets were prepared by repeating the same procedure as that of Example 1 except that the coating weights of the inner and outer coats were varied as specified in Table 1.

25 Table 1

Sample No.	Coating wt. of Eudragit E (inner coat)	Coating wt. of Eudragit S (outer coat)
30 ①	5%	20%
②	20%	20%
③	40%	20%
35 ④	20%	5%
⑤	20%	40%

\*: each percentage is given by weight and based on the solid drug

40 (c) In-vitro test on the influence of coating weight on the release of drug

The tablets (Sample Nos. ① to ⑤) were tested for the release of ketoprofen. Each tablet was incubated in a buffer of pH7.4 for 4 hours and put in a cecal solution. The results of the test were given in Fig. 3, wherein ○ represents the result of Sample No. ①, ● represents the result of Sample No. ②, △ that of Sample No. ③, ▲ that of Sample No. ④ and ○ that of Sample No. ⑤.

45 As apparent from the results of Fig. 3, equivalent release patterns were exhibited even when the coating weight of Eudragit S (outer coat) was varied. On the other hand, when the coating weight of Eudragit E (inner coat) was varied, however, somewhat different release patterns were exhibited.

50

55

Example 3

Ketoprofen-containing granules were prepared according to the following formula:

5	ketoprofen	20.0 pt. by wt.
	lactose	25.0 pt. by wt.
10	corn starch	30.0 pt. by wt.
	crystalline cellulose	20.0 pt. by wt.
	hydroxypropylcellulose	5.0 pt. by wt.

Ketoprofen, lactose, corn starch and crystalline cellulose were together mixed homogeneously, followed by the addition of a solution of hydroxypropylcellulose in purified water. The obtained mixture was kneaded and granulated with an extrusion granulator having a cylinder bore of 1 mm. The obtained granules were 15 dried and subjected to size classification. The resulting granules were coated as follows to form an inner coat. The weight increase was 5 mg per 100 mg of the granules.

20	Eudragit E (trade name)	7.0 pt. by wt.
	ethanol	70.0 pt. by wt.
	water	19.5 pt. by wt.
	talc	3.5 pt. by wt.

A solution comprising the above components was continuously sprayed on the granules preheated to 25 50 °C by the use of a flow coater at room temperature.

After the completion of the spraying, the resulting granules were dried and further coated with a solution comprising the following components to form an outer coat. The weight increase was 10 mg per 100 mg of the granules.

30	Eudragit L (trade name)	7.0 pt. by wt.
	ethanol	70.0 pt. by wt.
	water	18.8 pt. by wt.
	talc	3.5 pt. by wt.
35	polyethylene glycol 6000	0.7 pt. by wt.

(D) Test on the influence of pH on the release of ketoprofen

40 This test was conducted in the same manner as that of Example 1. The results are given in Fig. 4, wherein O represents the release pattern at pH2.0, ● represents that at pH7.4, and Δ represents that exhibited when the granules were allowed to stand in a buffer of pH7.4 for 4 hours and put in a cecal solution.

45 As apparent from the results of Fig. 4, no ketoprofen was released at pH2.0 or pH7.4, while specific release of ketoprofen was observed only in a cecal solution.

Example 4

A solid drug was prepared according to the following formula wherein fluorescein isothiocyanate dextran 50 (hereinafter abbreviated to "FITC dextran") was used as a model compound having dissolution characteristics different from those of ketoprofen.

55	FITC dextran	2.5 pt. by wt.
	magnesium stearate	1.0 pt. by wt.
	lactose	96.4 pt. by wt.
	carboxymethylcellulose	0.1 pt. by wt.

Both inner and outer coats were formed on the solid drug in the same manner as that of Example 1.

(E) Test on the influence of pH on the release of FITC dextran

5 This test was conducted in the same manner as that of Example 1. The results are given in Fig. 5, wherein  $\circ$  represents the release pattern at pH2.0,  $\bullet$  that at pH7.4, and  $\blacktriangle$  represents that exhibited when the preparation was allowed to stand in a buffer of pH7.4 for 4 hours and put in a cecal solution.

As apparent from the results of Fig. 5, little FITC dextran was released in a buffer of pH2.0 or pH7.4, while rapid release of FITC dextran was observed in a cecal solution.

10

Example 5

A solid drug having a calcitonin-containing core was prepared according to the following formula:

15

calcitonin	4.0 pt. by wt.
sucrose ester of fatty acid (F-160)	5.0 pt. by wt.
magnesium stearate	1.0 pt. by wt.
lactose	90.0 pt. by wt.

20

Calcitonin, lactose, magnesium stearate and sucrose ester of fatty acid were mixed together to form a homogeneous mixture. This mixture was compressed with a tablet machine into tablets having a diameter of 2 mm and a weight of 7 mg. The tablets were coated with a mixture comprising magnesium stearate (1 part by weight) and lactose (99 parts by weight). Both inner and outer coats were formed on the resulting tablets 25 in the same manner as that of Example 1.

(F) Test on the influence of pH on the release of calcitonin

This test was conducted in the same manner as that of Example 1. The results are given in Fig. 6, 30 wherein  $\circ$  represents the release pattern at pH2.0,  $\bullet$  represents that at pH7.4, and  $\blacktriangle$  represents that exhibited when the tablet was allowed to stand in a buffer of pH7.4 for 4 hours and put in a cecal solution.

As apparent from the results of Fig. 6, little calcitonin was released in a buffer of pH2.0 or pH7.4, while rapid release of calcitonin was observed in a cecal solution.

35

(G) Blood calcium concentration test using dog

In a similar manner to that of Example 1, the tablet prepared in Example 5 was administered to dogs to determine the blood calcium concentration. The results are given in Fig. 7, wherein  $\bullet$  represents the result exhibited when the tablet of Example 5 was administered, and  $\circ$  represents the result exhibited when the tablet of Example 5 was not coated at all.

As apparent from the results of Fig. 7, little lowering was observed in the blood calcium concentration, when the uncoated tablet was administered. On the other hand, lowering was observed in the blood calcium concentration 10 hours after the administration and later, when the double-coated tablet was administered.

45

Example 6

Five kinds of tablets were prepared in a similar manner to that of Example 1 except that the coating weights of the inner and outer coats were varied as specified in Table 2.

50

55

Table 2

Sample No.	Coating wt. of Eudragit E (inner coat)	Coating wt. of Eudragit S (outer coat)
①	3%	20%
②	5%	20%
③	7%	20%
④	5%	5%
⑤	5%	3%

\*: each percentage is given by weight and based on the solid drug

## (H) In vitro test on the influence of coating weight on the release of drug

These tablets were tested for the release of ketoprofen. Each tablet was incubated in a buffer of pH7.4 for 4 hours and put in a cecal solution. The results are given in Fig. 8, wherein ○ represents the result of Sample No. ①, ● represents that of Sample No. ②, △ represents that of Sample No. ③, ▲ represents that of Sample No. ④, and □ represents that of Sample No. ⑤.

As apparent from the results of Fig. 8, equivalent release patterns were exhibited, even when the coating weight of Eudragit S (outer coat) was varied. On the other hand, when the coating weight of Eudragit E (inner coat) was 5% by weight or below, excellent release patterns were exhibited.

Example 7

Calcitonin-containing solid drugs were prepared according to the following formulae:

Table 3

Sample No.	①	②	③	④	⑤
calcitonin	0.28 mg				
sucrose ester of fatty acid (F-160)	25%				
sodium caprate		25%		25%	
dipotassium glycyrrhizinate			25%	25%	
magnesium stearate	1%	1%	1%	1%	1%
lactose	74.0%	74.0%	74.0%	49.0%	99.0%

\* each percentage is given by weight

The components of each formula were mixed together to form a homogeneous mixture. This mixture was compressed into tablets having a diameter of 7 mm and a weight of 200 mg with a tablet machine. These tablets were each coated as follows.

Eudragit E (trade name)	7.0 pt. by wt.
ethanol	70.0 pt. by wt.
water	19.5 pt. by wt.
talc	3.5 pt. by wt.

A solution comprising the above components was continuously sprayed at 50 °C on the tablet preheated to 50 °C. The weight increase of the solid drug was 14 mg. The resulting tablets were dried and further coated as follows.

5

Eudragit S	7.0 pt. by wt.
ethanol	70.0 pt. by wt.
water	18.8 pt. by wt.
talc	3.5 pt. by wt.
polyethylene glycol 6000	0.7 pt. by wt.

A solution comprising the above components was continuously sprayed at 50 °C on the tablets preheated to 50 °C. The weight increase of the solid drug was 14 mg.

10

(I) In vivo test on the effect of sorbefacient

In a similar manner to that of Example 5, the tablets prepared in Example 7 were each administered to dogs to determine the blood calcium concentrations. The results are given in Fig. 9, wherein ○ represents the result of Sample No. ①, ○ represents that of Sample No. ②, △ represents that of Sample No. ③, △ represents that of Sample No. ④, and □ represents that of Sample No. ⑤.

As apparent from the results of Fig. 9, lowering in the blood calcium concentration was observed 10 hours after the administration and later when Sample No. ①, ②, ③ or ④ was administered, while no lowering was observed when Sample No. ⑤ not containing any sorbefacient was administered. Particularly excellent lowering in the blood calcium concentration was observed when sodium caprate or dipotassium glycyrrhizinate was used as the sorbefacient.

Industrial Applicability

25 The oral pharmaceutical preparation releasable in the lower digestive tract according to the present invention is characterized by the specific release of the active ingredient therefrom caused by the unique pH change in the digestive tract, so that the preparation little causes lowering or loss in the availability and scattering among individuals, though the preparation targetting the lower digestive tract (such as the large intestine) according to the prior art suffered from these disadvantageous phenomena.

30 In particular, the oral preparation of the present invention passes through the stomach and small intestine without being affected, and is disintegrated for the first time or arrival at the large intestine (i.e., the lower digestive tract) to release the active ingredient, so that the active ingredient is effectively absorbed in the large intestine to give a remarkably enhanced therapeutic effect and other excellent effects. Accordingly, the preparation of the present invention is useful as a pharmaceutical preparation for oral administration for treating a patient with a colonic or large intestine disease.

Claims

1. An oral pharmaceutical preparation releasable in the lower digestive tract, characterized by having a double-coated structure wherein a solid drug having a core containing an active ingredient is covered with both an inner coat made of a cationic polymer and an outer coat made of an anionic polymer.
2. An oral pharmaceutical preparation releasable in the lower digestive tract as set forth in claim 1, wherein the core contains a sorbefacient.
3. An oral pharmaceutical preparation releasable in the lower digestive tract as set forth in claim 2, wherein the sorbefacient is selected from among sugar ester, sucrose esters of fatty acid, glycyllsinate salts, glycyrrhetic acid, dipotassium glycyrrhizinate, bile acid, glycerol esters of fatty acid, 1-[(2-(decylthio)ethyl]azacyclopentan-2-one, adipic acid, basic amino acids, polyethylene glycol and sodium caprate.
4. An oral pharmaceutical preparation releasable in the lower digestive tract as set forth in claim 1, wherein the cationic polymer is an aminoalkyl methacrylate copolymer or polyvinyl acetal diethylaminoacetate which is soluble or swelling at a pH of 6.0 or below.
5. An oral pharmaceutical preparation releasable in the lower digestive tract as set forth in claim 1, wherein the anionic polymer is selected from the group consisting of methacrylic acid copolymer L, methacrylic acid copolymer S, hydroxypropylmethylcellulose, hydroxyethylcellulose phthalate, polyvinyl

acetate phthalate, cellulose acetate and cellulose acetate tetrahydrophthalate which are soluble at a pH of 5.5 or above.

6. An oral pharmaceutical preparation releasable in the lower digestive tract as set forth in claim 1, which  
5 is disintegrated and absorbed in the lower digestive tract wherein the pH changes to below the range of  
7.0 to 5.5.
7. An oral pharmaceutical preparation releasable in the lower digestive tract as set forth in claim 1,  
10 wherein the active ingredient is selected from the group consisting of peptides, proteins, anti-  
inflammatory agents, antineoplastic agents, antibiotics, and chemotherapeutics.

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FIG. 1

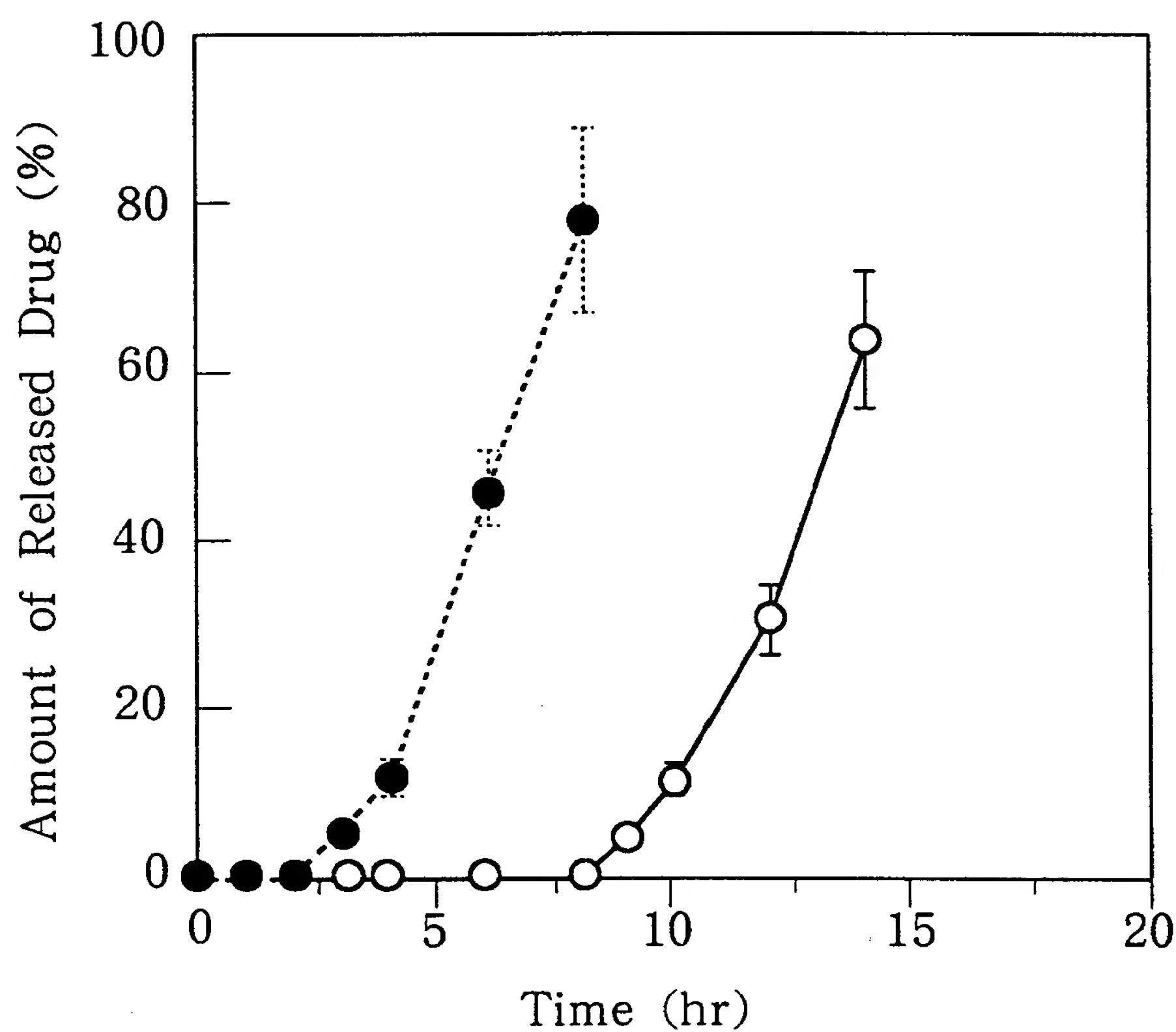


FIG. 2

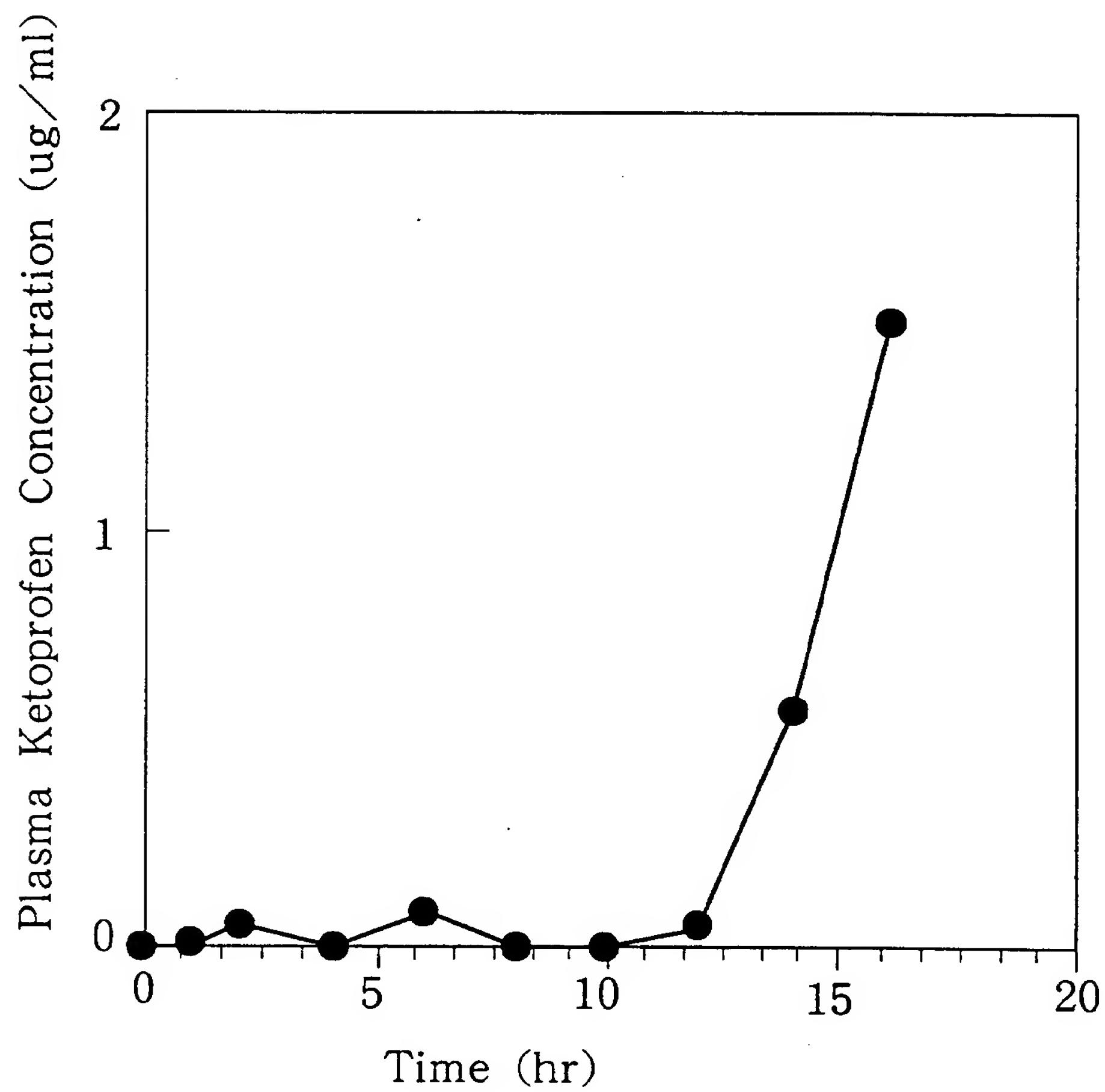


FIG. 3

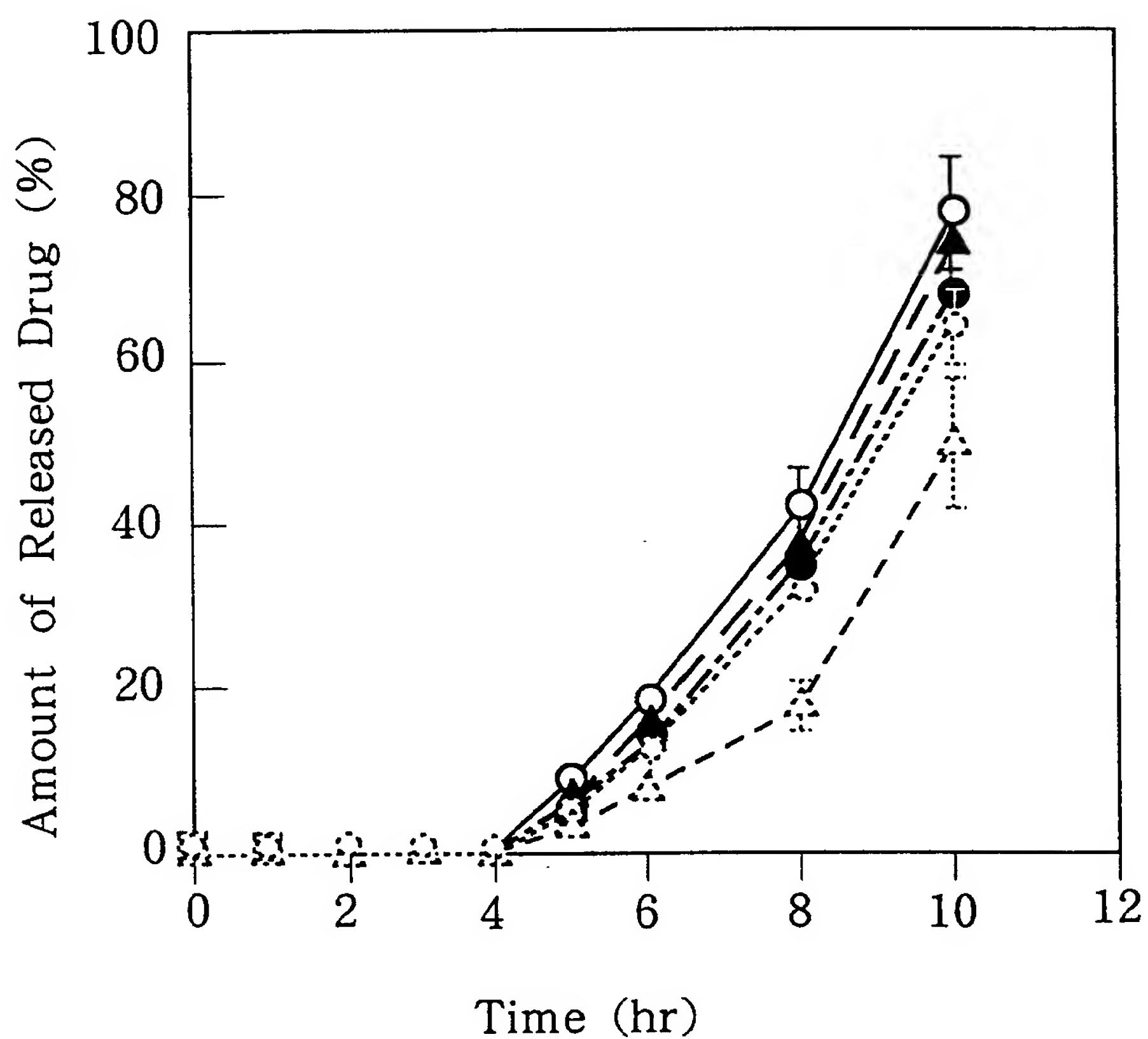


FIG. 4

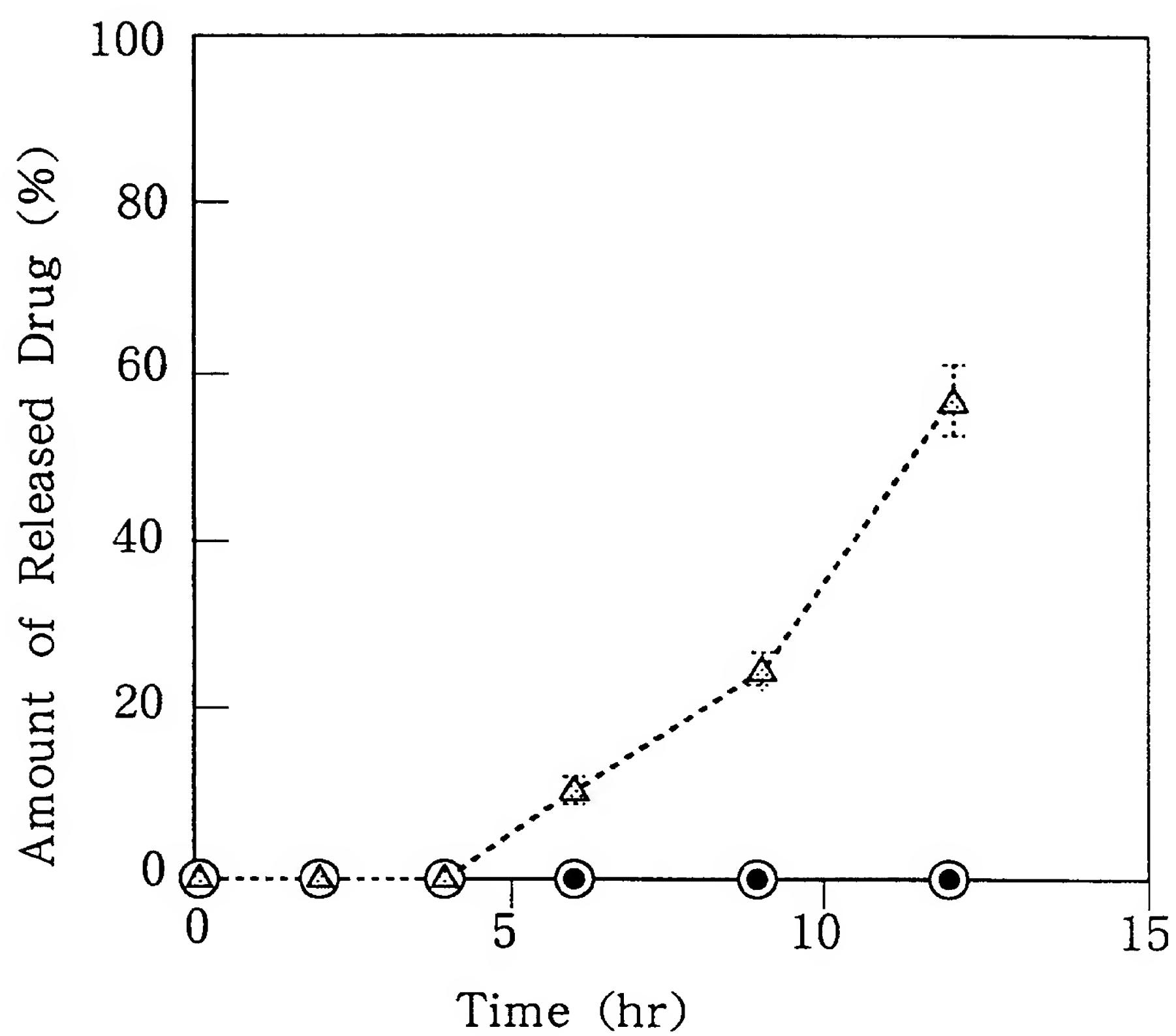


FIG. 5

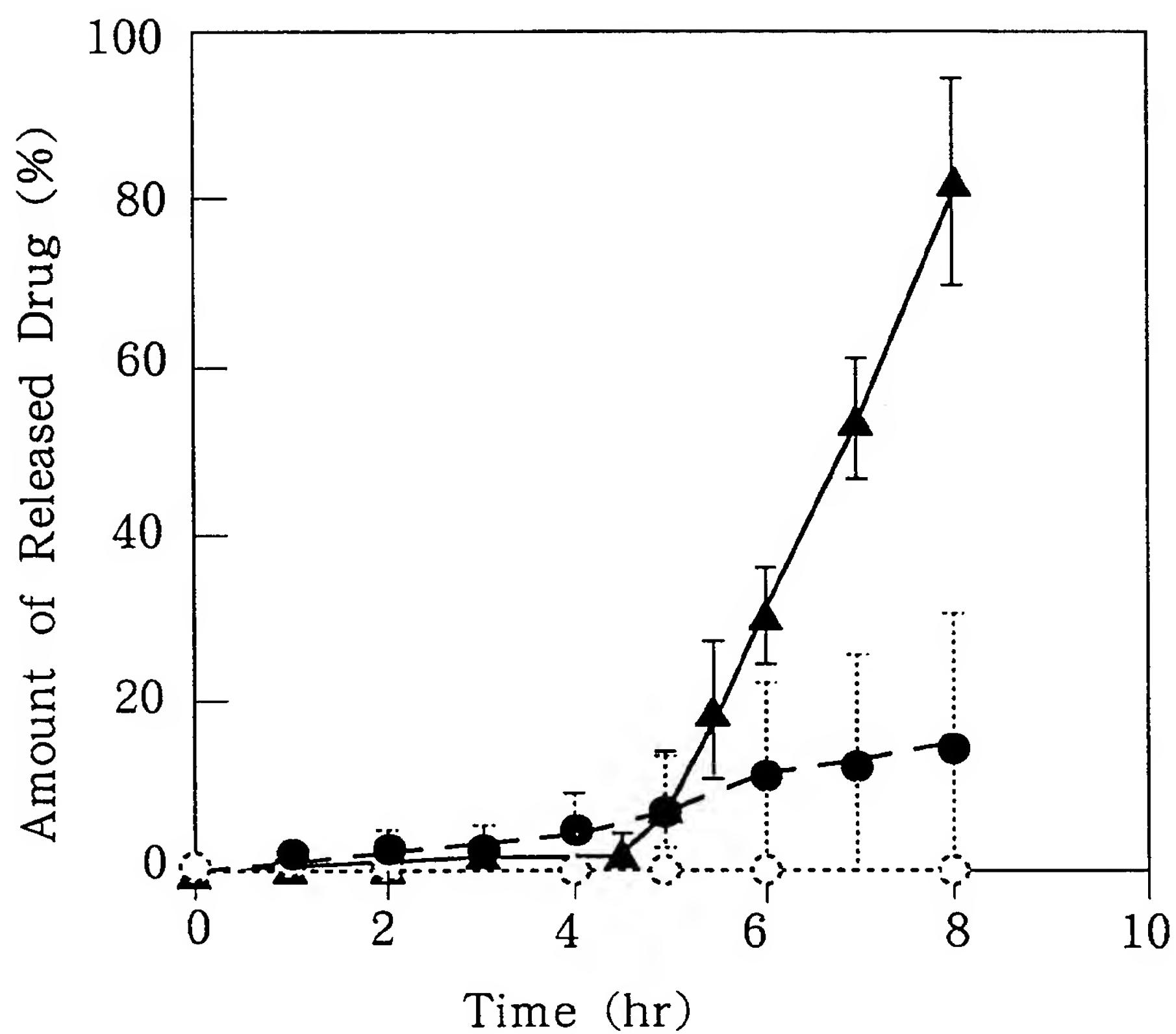


FIG. 6

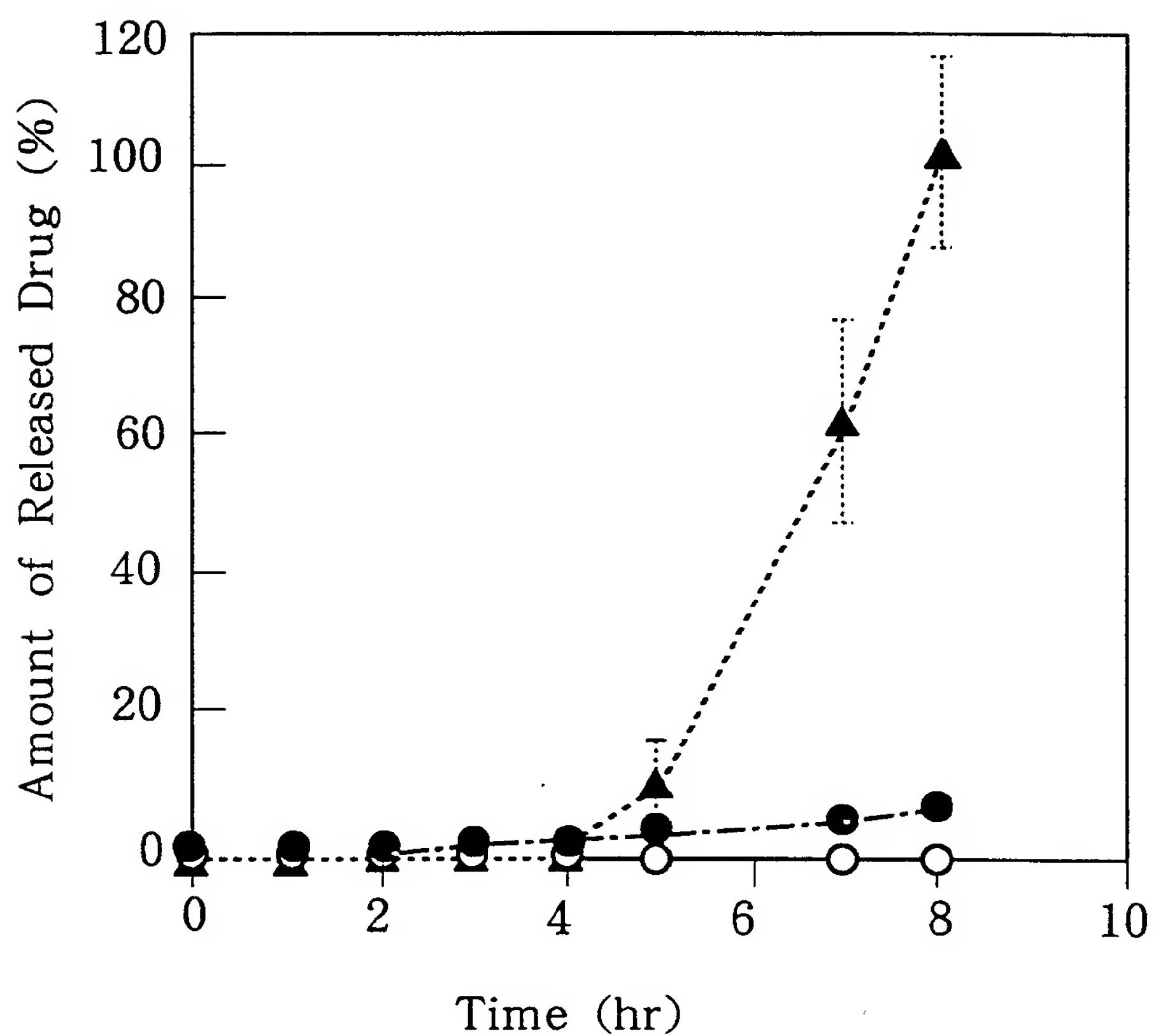


FIG. 7

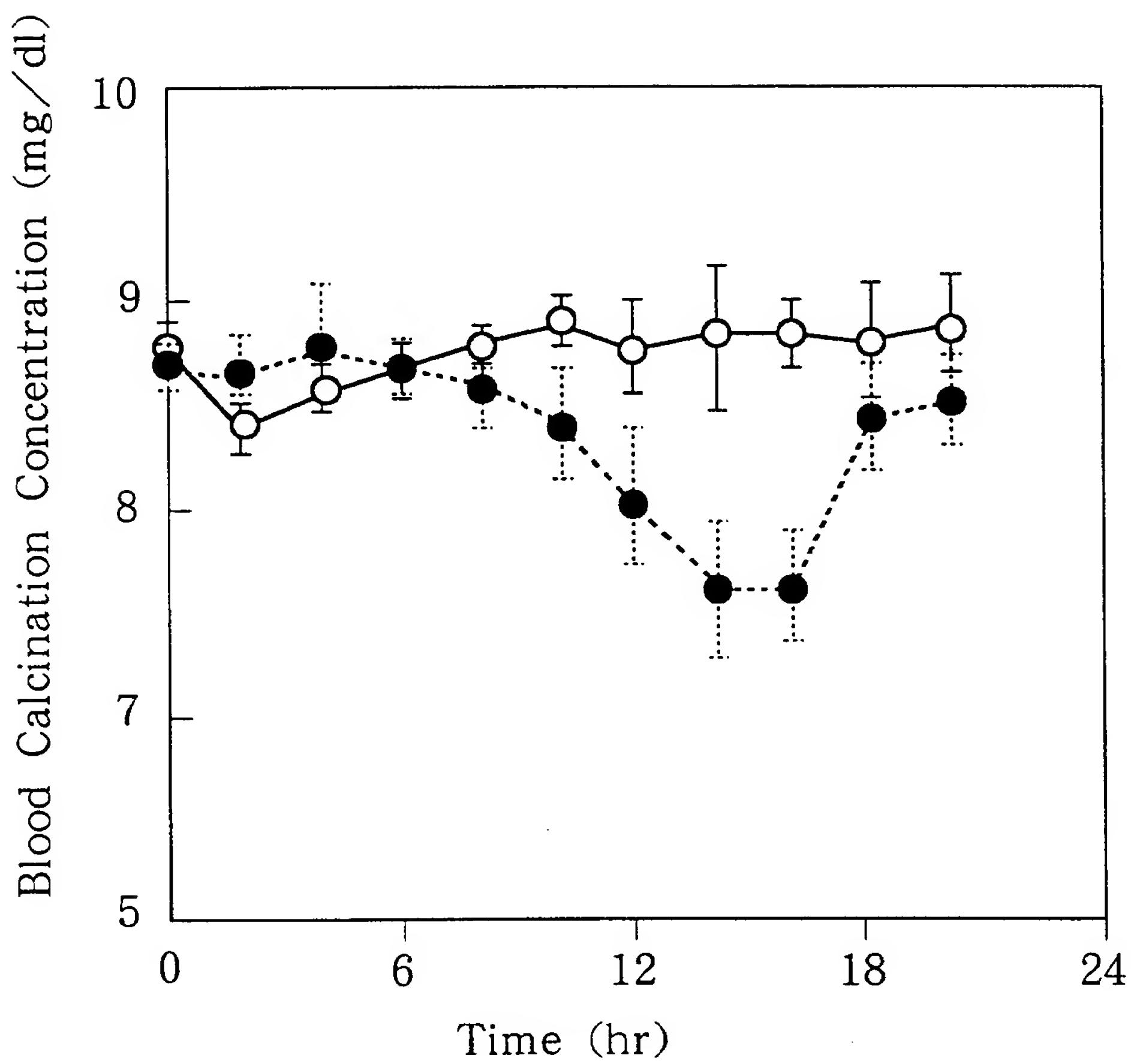


FIG. 8

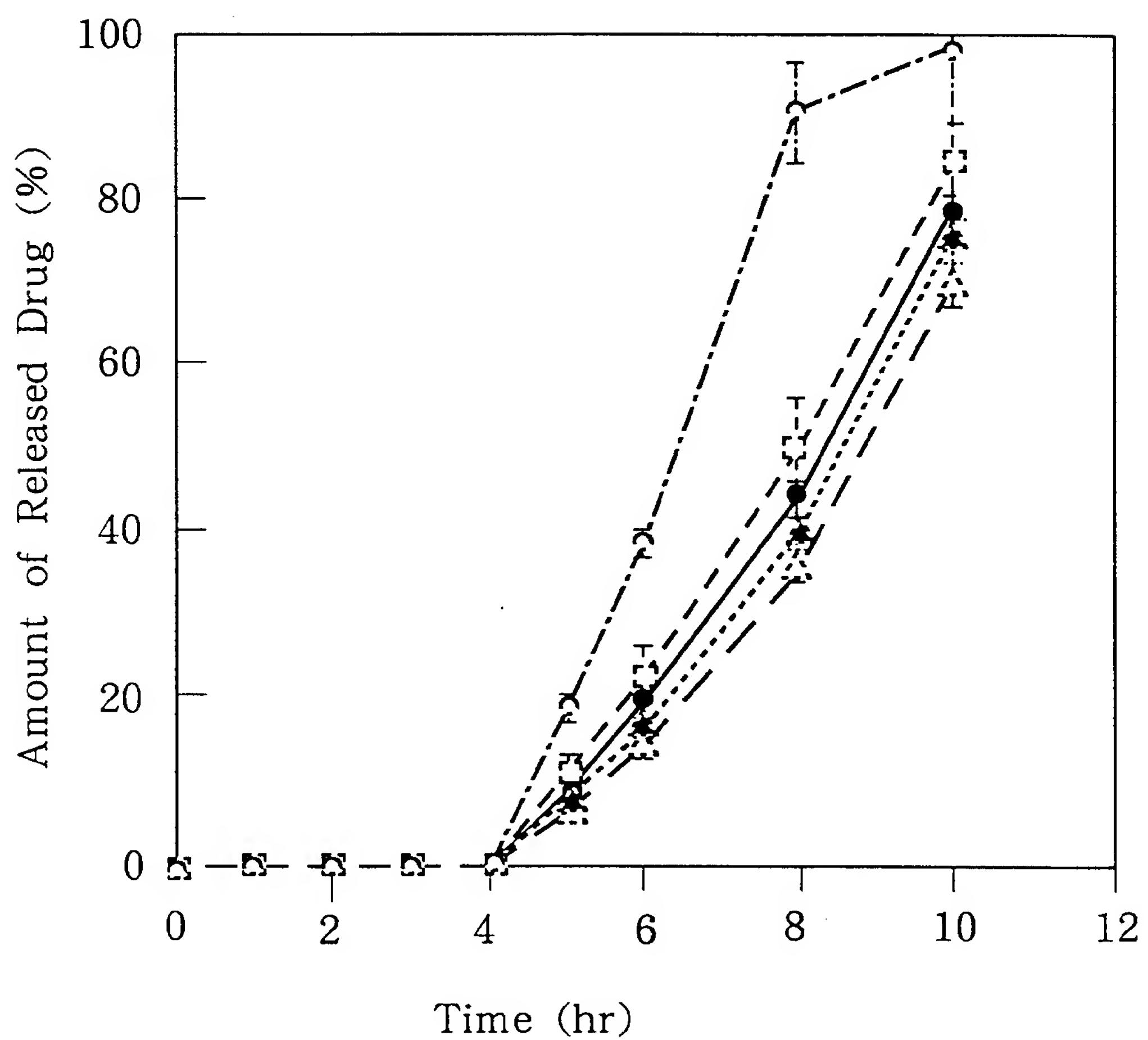
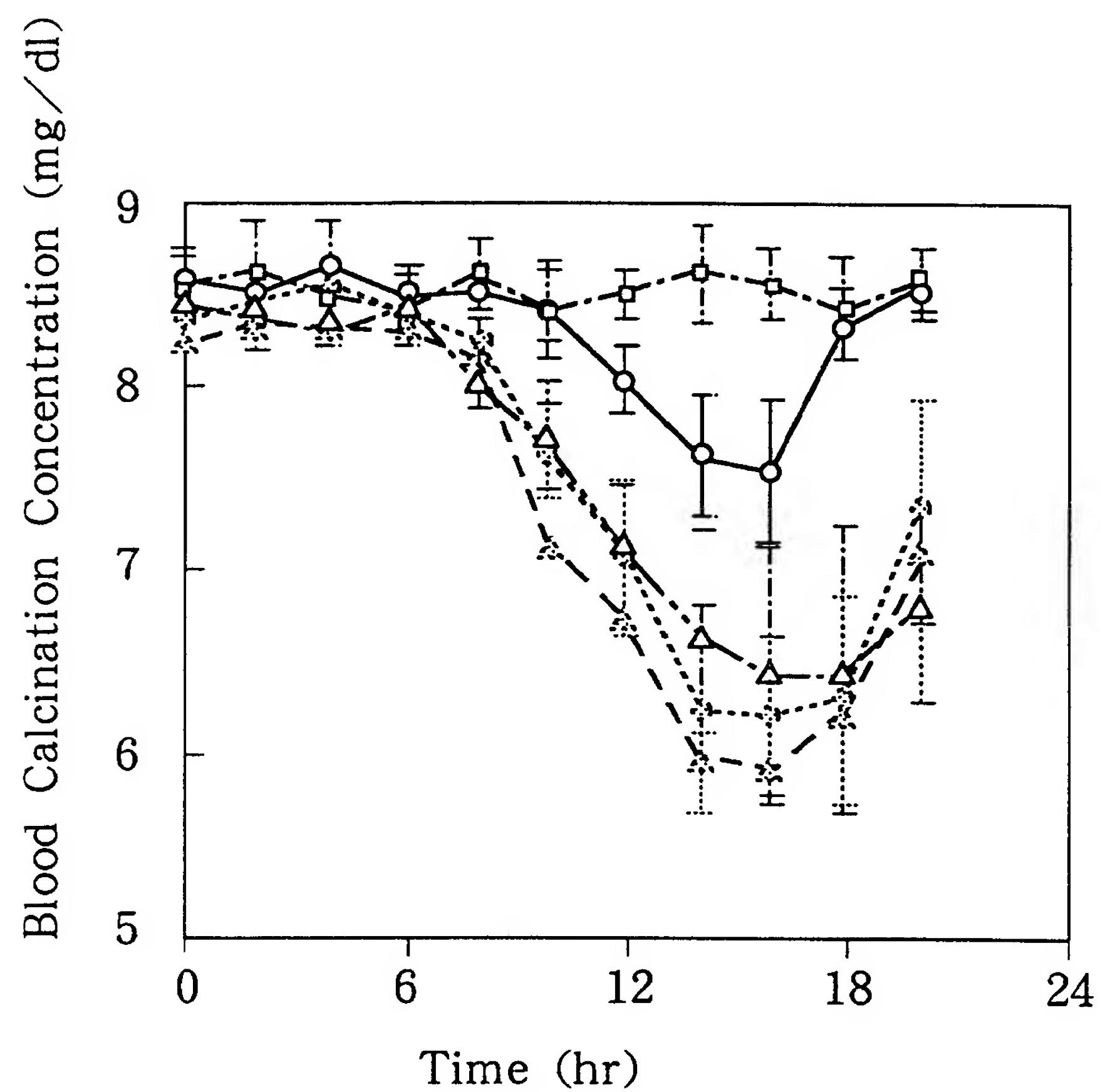


FIG. 9



**INTERNATIONAL SEARCH REPORT**

International application No.

PCT/JP93/01608

**A. CLASSIFICATION OF SUBJECT MATTER**

Int. Cl<sup>5</sup> A61K9/30

According to International Patent Classification (IPC) or to both national classification and IPC

**B. FIELDS SEARCHED**

Minimum documentation searched (classification system followed by classification symbols)

Int. Cl<sup>5</sup> A61K9/20-9/36, A61K47/32, 47/36-47/38

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

**C. DOCUMENTS CONSIDERED TO BE RELEVANT**

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	JP, A, 58-501174 (J. B. Tirott Ltd.), July 21, 1983 (21. 07. 83) & WO, A1, 83/00435	1-7
A	JP, A, 4-501411 (Beringer Ingerheim Italia S.p.A.), March 12, 1992 (12. 03. 92) & WO, A1, 90/04386	1-7
A	JP, A, 4-225922 (Teikoku Seiyaku Co., Ltd. and another), August 14, 1992 (14. 08. 92), (Family: none)	1-7

<input type="checkbox"/>	Further documents are listed in the continuation of Box C.	<input type="checkbox"/>	See patent family annex.
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"A" document defining the general state of the art which is not considered to be of particular relevance	"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention		
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"O" document referring to an oral disclosure, use, exhibition or other means	"&" document member of the same patent family		
"P" document published prior to the international filing date but later than the priority date claimed			

Date of the actual completion of the international search January 24, 1994 (24. 01. 94)	Date of mailing of the international search report February 15, 1994 (15. 02. 94)
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Name and mailing address of the ISA/ Japanese Patent Office	Authorized officer
Facsimile No.	Telephone No.